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and formulas of acids are based on an acid's consisting of an anion and enough hydrogen ions to make the acid electrically neutral. The general formula for an acid is $H_n X$, where X is an anion, H is a hydrogen ion, and n is the number of hydrogen ions needed to make the acid neutral.

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SECTION 9.4 NAMING AND WRITING

FORMULAS FOR ACIDS AND BASES

(pages 271-273) This section explains

the three rules for naming acids and

shows how these rules can also be used

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to write the formulas for acids. <https://www.foothillfalcons.org/ourpages/auto/2010/10/7/54585836/Chap9WBKEY.pdf>
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to answer the following questions

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