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Sokollm (In order of which they appear)
Key Concepts: Terms in this set (20)
Heat. the transfer of thermal energy
from one object to another because of a
difference in temperature. Temperature.
a measure of how hot or cold an ...

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Summary 16.1 Thermal Energy and
Matter Heat flows spontaneously from
hot objects to cold objects. • Heat is the
transfer of thermal energy from one
object to another because of a
temperature difference. Temperature is
related to the average kinetic energy of

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the particles in

Chapter 16 Thermal Energy and Heat

Physical Science Chapter 16: Thermal Energy and Heat. Heat is the transfer of thermal energy from one object to another as the result of a difference in _____ . _____ produces

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Physical Science Chapter 16: Thermal Energy and Heat ...

It states that thermal energy can flow from colder objects to hotter objects only if work is done on the system. Third law of thermodynamics It states that absolute zero cannot be reached.

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energy from one object to another because of a temperature difference.

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is a measure of how hot or cold an object is compared to a ref.... Increase in volume of material when its temperature increases. Conduction. is the transfer of thermal energy through touching with no ove.... Heat Engine. is any device that converts heat into work. 36 Terms. etakp. Chapter 16 Thermal Energy and

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16.1 - Thermal Energy and Matter . Work
and Heat Heat is the transfer of thermal
energy from one object to another
because of temperature differences Heat

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flows spontaneously from hot objects to cold objects Imagine two glasses with differing amounts of water in them at the same temperature The glass with ...

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16.1 Thermal Energy & Matter. Work and Heat. Heat -the transfer of thermal energy from one object to another because of a temperature difference Heat flows from higher temps to lower

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temps. Temperature is related to the kinetic energy of the particles: particles move around as they heat

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Tools. Copy this to my account; E-mail to a friend; Find other activities; Start over; Help; A B; heat: the transfer of thermal

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energy from one object to another
because of a difference in temperature:
temperature: a measure of how hot or
cold an object is compared to a
reference point:

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Section 16.1 Matter

Section 16.1 Thermal Energy and Matter (pages 474–478) This section defines heat and describes how work, temperature, and thermal energy are related to heat Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained Reading Strategy (page 474)

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Calculating with Specific Heat How much heat is required to raise the temperature of a gold earring from 25.00C to 30.00C? The earring weighs 25 grams, and the specific heat of gold is 0.128 J/ge0C. 1. 2. 3. Read and Understand What information are you given? Specific heat = $c = 0.128 \text{ J/ g. }^{\circ}\text{C}$ Mass = $m = 25.0$

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another as the result of a difference in
temperature. True. T/F: On the Celsius
Scale, the reference points for

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temperature are the freezing and boiling points of water. thermal energy ...

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